

Composition of atmosphere air by volume:

Nitrogen ( $N_2$ )  $\rightarrow$  78%

Oxygen ( $O_2$ )  $\rightarrow$  21%

Argon  $\rightarrow$  0.9%

Other gases  $\rightarrow$  0.1% ( $CO_2, CH_4, He, Ne$  etc.)

$CH_4 \rightarrow$  coal in mines and Geo-thermal produce.

$\rightarrow$  Argon is a noble gas which is high amount of gas present in atmosphere.

Damp means vapour or smoke

Damp is divided into 5 types such as -

1) Black Damp: Suffocating Damp ( $CO_2$ )

$\rightarrow$  It is mainly known as  $CO_2$  because it creates difficulties to breathing. ~~etc~~ (choke damp)

$\rightarrow$  It also extinguishing fire.

2) Fire Damp:

$\rightarrow$  These are inflammable gas which catches fire with the slight rise of temperature

$CH_4, C_2H_6, C_3H_8, C_4H_{10}$  etc.

3) White Damp:

$\rightarrow$  It is other name known as carbon monoxide ( $CO$ ).

$\rightarrow$   $CO$  has the higher affinity to the haemoglobin of our blood.

4) Stink Damp: ( $H_2S$ )  $\rightarrow$  Hydrogen sulphate

$\rightarrow$  The stink means unpleasant smell.

ex -  $H_2S$  which is like rotten egg.

5) After Damp:

→ It is the gas exist after the explosion of fire damp / coal dust.

→ There are many gas present in our mine's atmosphere. such as i)  $\text{CO}_2$  ii)  $\text{O}_2$   
iii)  $\text{CH}_4$  iv)  $\text{H}_2\text{S}$  v)  $\text{CO}$

→ The physical properties of mine gases.

i)  $\text{CO}_2$

→ Specific gravity - 1.529

→ colorless

→ odorless

→ Heavier than air

→ Acidic taste at high concentrations

ii)  $\text{CO}$

→ Specific gravity - 0.967

→ flammable

→ colorless

→ tasteless

→ odorless

→ lighter than the air

iii)  $\text{H}_2\text{S}$  :-

→ Specific gravity - 1.191

→ colorless

→ extremely flammable

→ Heavier than air

→ highly soluble

→ odor to similar to rotten eggs at very low concentrations (0.003) ppm.

wt O<sub>2</sub> →

- Specific gravity → 1.105
- colorless
- Odorless
- Tasteless
- Non-explosive
- Supports combustion
- Heavier than air
- easily displaced by other gases introduced into the atmosphere.

wt CH<sub>4</sub> →

- Specific gravity → 0.5545
- colorless
- odorless
- Tasteless
- flammable
- lighter than air
- largest component of fire damp (70 to 98%)

Flame safety Lamp →

→ To get rid of recurring problem of fire damp explosion due to open lamp considering this problem Sir. Humphry Davy in 1815 came forward with a revolutionary device known as Davy safety lamp. The flame of the lamp was covered by a wire gauze of copper having good thermal conductivity. The flame heat was <sup>dissipated</sup> ~~dissipated~~ (waste) through the wire gauze thus not enough of heat pass to ignite the explosive mixture surrounding

to the lamp. The methane entire along with inlet air was simply burnt off.

### Different parts of safety lamp:

- 1) carrying hook
- 2) Vent holes
- 3) Bonnet
- 4) Pillar
- 5) magnetic lock
- 6) Inner gauge ~~and~~ outer gauge
- 7) outer gauge
- 8) Borosil glass
- 9) Asbestos ring
- 10) wick retainer holder
- 11) wick
- 12) Regulator of wick / wick asbestos
- 13) Oil vessel

#### 1) carrying hook:-

→ For holding the lamp and carrying it at knee height.

#### 2) vent holes:-

→ The combustion gas ~~both~~ goes into lamp through this holes.

#### 3) Bonnet:-

→ It is used for taking care

→ It protects to the wire gauge from damage due to external ~~heat~~ hit.

It also guards the flame against high air velocity.

4) Pillars :-

→ The bonnet rest of on pillars over the oil vessel generally 5 pillars are provided protect to the glass.

5) magnetic lock :-

→ To unlock the lamp the iron Pt block fitted to the spring is brought into a contact a powerful magnet so that due to contraction of the spring the pointer is groove.

6) Outer and inner gauze :-

→ The main function of two gauzes is to prevent the passage of flame from interior lamp for the safety against explosion of surrounding atmosphere even it contains explosive mixture. The gauze are made of copper wire of 28 mesh. While hot combustion gas passes through the part of the heat is absorbed by gauzes which than ~~dissipate~~ dissipate heat by way of radiation. Thus the temperature of combustion gas ~~is~~ leaving the gauze as well as outer gauze temperature below the ignition or fire damp.

7) ~~Asbestos~~ Borosil glass :-

→ It is blend of silica and Boron which <sup>mix</sup> makes it reinforced / strong glass.

8) Asbestos ring:-

→ We use two asbestos rings, one is at the bottom and top of the outer glass are put to make the joint like proof and prevent the explosion flame gas come outside.

9) Wick & wick holder:-

→ Wick emits the active flame when set out from fire in it. The container which holds the wick is known as wick holder.

10) Wick asbestos:-

→ The pointer passes through a hollow cylinder welded in the oil vessel. The flame height can be regulated by pointer without opening the lamp.

Oil use for lamp:-

(i) Kerosene for Volux GL-5, non-relighting type flame safety flame lamp.

(ii) Esso - solvent sprit no. 1425 and Burtman Shell BP-SS for Volux GL-60 and GL-70 relighting type safety lamp.

Exudation:-

→ Exudation, gradual discharge of any gas especially fire damp from the coal seam due to huge accumulation of gases inside the gap between two strata or forces ~~to~~ <sup>with</sup> the mineral material.

Factor's affecting on spontaneous heating:-i) Chemical composition of coal:-

High moisture and high volatile susceptible to spontaneous heating. All bright coal which

25% or more volatile matter and 7-15% of moisture are prone to spontaneous heating

~~moisture~~

carbon:-

→ High percentage of carbon content of a coal is not conducive (helpful) to S.H. whereas the coal contains less percentage of carbon are conducive to spontaneous heating

ii) Bonded oxygen:-

→ Prone to spontaneous heating of coal decreases with decreasing oxygen content in the volatile matter of coal

iii) Bonded constituents of coal:-

→ The bright bands of coal i.e. vitrinite, clarinite are more prone to spontaneous heating than the dull constituents i.e. durinite and fusinite.

iii) Friability:-

coal which is easily crossed crushed and broken into smaller size is more prone to spontaneous heating than hard coal.

iv) Presence of Iron Pyrate:-

coal containing iron pyrate is more prone to spontaneous heating.

Symptoms of Spontaneous heating:-

- 1) smell and resembles burning of timber
- 2) Increasing in Wet Bulb temperature and dry bulb temperature.
- 3) water droplets on the surface of metal and timber.
- 4) Uneasiness in breathing

Note:- Grinhm's ratio:-

→ It is the ratio of carbon monoxide (CO) to the oxygen ( $O_2$ ) in a gassy underground mine.

→ which shows whether workable condition or not.

Incubation period:-

→ The time ~~elapsed~~ <sup>elapsed</sup> between ~~the~~ when coal seam is subjected to S.H and the appearance of <sup>active</sup> force <sub>on coal seam</sub> due to ignition of coal ~~seam~~ <sup>as</sup> ~~the~~ <sup>critical</sup> ~~rise of~~ <sup>temperature</sup> it reaches its critical temperature.



Q4 In atmosphere contain

$$N_2 = 79.04\%$$

$$O_2 = 20.93\%$$

$$CO_2 = 0.03\%$$

$$N_2 = 78.72\%$$

$$O_2 = 19.95\%$$

$$CO_2 = 0.39\%$$

$$CO = 0.005\%$$

In

$$79.04\% N_2 = 20.93\% O_2$$

$$\frac{1}{79.04\%} N_2 = \frac{20.93\%}{79.04\%} \star$$

$$78.72\% N_2 = \frac{20.93\%}{79.04\%} \times 78.72\%$$

$$= 20.85\% O_2$$

$$O_2 \text{ absorb} = 20.85 - 19.95$$

$$= 0.9\%$$

$$CO_2 = 0.39\% - 0.03\%$$

$$= 0.36\%$$

$$\text{In stream ratio} = \frac{CO_2}{O_2} = \frac{0.36}{0.9} \times 100 = 40\%$$

$$\text{of } \frac{CO}{O_2} = \frac{0.005}{0.9} \times 100 = 0.56\%$$

$$Q5 \quad N_2 = 77.57\%$$

$$O_2 = 18.56\%$$

$$CO_2 = 0.67\%$$

$$CO = 0.09$$

$$N_2 = 79.04\%$$

$$O_2 = 20.93\%$$

$$CO_2 = 0.03\%$$

$$79.04\% N_2 = 20.93\% O_2$$

$$\frac{1}{79.04} N_2 = \frac{20.93\%}{79.04}$$

$$77.57\% N_2 = \frac{20.93\%}{79.04} \times 77.57\%$$

$$= 20.54\% O_2$$

$$O_2 = 20.54 - 18.56$$

$$= 1.98 \%$$

$$CO_2 = 0.67 - 0.03$$

$$= 0.64$$

In gram ratio =  $\frac{CO_2}{O_2} = \frac{0.64}{1.98} \times 100 = 32.32\%$

or  $\frac{CO}{O_2} = \frac{0.09}{1.98} \times 100 = 4.54\%$

Rf  $N_2 = 79.04\%$

$O_2 = 20.93\%$

$CO_2 = 0.03\%$

$N_2 = 76.65\%$

$O_2 = 17.54\%$

$CO_2 = 0.45\%$

$CO = 0.006\%$

$79.04 N_2 = 20.93 O_2$

$1 N_2 = \frac{20.93}{79.04}$

$76.65 N_2 = \frac{20.93}{79.04} \times 76.65$

$= 20.29 O_2$

$O_2 \text{ absorb} = 20.29 - 17.54$

$= 2.75\% O_2$

$CO_2 = 0.45 - 0.03 = 0.42$

In gram ratio =  $\frac{CO_2}{O_2} = \frac{0.42}{2.75} \times 100 = 15.27$

or  $\frac{CO}{O_2} = \frac{0.006}{2.75} \times 100$

$= 0.21\%$

## Mine Explosion

→ It due to Ignition of fire damp and coal dust  
Explosion:-

It is sudden or out burst or combustion process of great intensity which release large amount of energy and gaseous product.

### Limit of Flammability:-

→ methane ( $\text{CH}_4$ ) → lower flammability limit 5.4%.

### Limit of flammability mixture:-

→ For a mixture of inflammable gases the lower and upper flammability limit can be calculated

by A = Graham's law

B = Dalton's law

C = Boyle's law

D = Le-Chatelier law

### Graham's law:-

It is state that of diffusion of gas is inversely proportional

$$\frac{\text{rate}_1}{\text{rate}_2} = \sqrt{\frac{m_1}{m_2}}$$

### Le-Chatelier law:-

It is used to find out the lower explosibility a combustible mixture of gas.

$$\frac{P_T}{P_{low}} = \frac{P_1}{L_1} + \frac{P_2}{L_2} + \frac{P_3}{L_3}$$

where  $P_T$  = Total Percentage of combustible gas

$P_{low}$  = lower explosibility of mixture

$P_1, P_2, P_3$  = concentration of individual gases in Percentage

$L_1, L_2, L_3$  = lower explosibility limit of individual gases

| at | conc.               | LEL                |
|----|---------------------|--------------------|
|    | $\text{CH}_4 = 8\%$ | $\text{CH}_4 = 5$  |
|    | $\text{CO} = 5\%$   | $\text{CO} = 12.5$ |
|    | $\text{H}_2 = 3\%$  | $\text{H}_2 = 4$   |

$$\Rightarrow \frac{P_T}{P_{\text{low}}} = \frac{P_1}{L_1} + \frac{P_2}{L_2} + \frac{P_3}{L_3}$$

$$\Rightarrow \frac{16}{P_{\text{low}}} = \frac{8}{5} + \frac{5}{12.5} + \frac{3}{4}$$

$$\Rightarrow \frac{16}{P_{\text{low}}} = 1.6 + 0.4 + 0.75$$

$$\Rightarrow \frac{16}{P_{\text{low}}} = 2.75$$

$$\Rightarrow P_{\text{low}} = \frac{16}{2.75}$$

$$\Rightarrow P_{\text{low}} = 5.81$$

NOTE: 1 mole of  $\text{CH}_4 =$  atomic mass  $\times$  A.M.  
 $= 12 + 2 \times 4$   
 $= 12 + 8$   
 $= 20$

Coal-dust explosion:

→ It is a sudden combustion process of grade intensity and it has use destructive effect through pressure and heat.

→ Ignition temperature of coal is  $600^\circ\text{C} - 700^\circ\text{C}$

Factors Affecting coal dust explosion:

→ particle size:

→ Size of the coal dust explosion:  $10 - 100 \mu\text{m}$  are very dangerous to explosion.

## MINE INUNDATION

### Introduction:-

→ Sudden leakage of water either from a surface, or water logged area or near by water body into active watering area of a mine.

### Measure against inundation:-

1) surface measure

2) underground measure

#### 1) surface measure:-

→ locating shaft away from water logged area

→ filling with hard material in abandoned shaft or borehole.

→ making diversion reaches.

→ concrete on the surface to channelize the surface underground.

→ Back filling of surface excavation

→ const. of dam and reserve

### water dam:-

→ It is a permanent artificial barrier build into the mine working to protect transmission.

### Types of dam:-

1) Flat dam

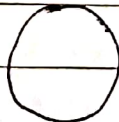


2) cylindrical dam / Arch

3) ~~curved~~



4) Spherical dam



### Factor considering to design a dam:

- size of the roadway
- nature of adjacent strata
- Estimated water pressure
- material used for construction
- shape of dam

### Thickness of dam:

#### Flat dam / Rectangular dam:

$$T = \frac{Pbh}{2(b+h)\tau_s}$$

where  $T$  = thickness of dam

$P$  = maximum water pressure

$b$  = width of dam

$h$  = height of dam

$\tau_s$  = shear strength of material

Q) Calculate the thickness of a flat dam to where  
 as width of dam 5m, height is 8m shearing  
 strength of material 0.15 MP.

ans:

$$P = \rho gh = 1000 \times 9.8 \times 8$$

$$P = 78400 \quad 1 \text{ MPa} = 10^6 \text{ p}$$

$$T = \frac{Pbh}{2(b+h)\tau_s}$$

$$= \frac{78400 \times 5 \times 8}{2(5+8) \times 0.15}$$

$$= \frac{78400 \times 5 \times 8}{2(5+8) \times 0.15}$$

$$= \frac{78400 \times 5 \times 8}{20 \times 0.15} = \frac{784000 \times 5 \times 8}{20 \times 1500 \times 100}$$

$$= \frac{3136000}{3900000} = 0.804$$

Q. Why Inundation occurs in mines?

- In accuracy old plans.
- The lake of old plans.
- error of judgement or neglect or precision
- on suspected presence of old shaft, bore hole or drift connecting to full of water.
- encroaching into the working of adjacent mines by crossing the common boundary when the state of working is unknown.
- sudden collapse of water bearing in strata due to faulty method of working.

Water management system:

- Revolution of water source
- Revolution of mine water needs
- Development of data collection plan and field investigation.
- water distribution and conveyance
- site specific water balance
- mine de-watering
- surface water management
- flood-plain analysis
- water reservoir and dams.

## Noise and vibration

### Noise :-

→ Noise is a any sound ~~that~~ <sup>whose</sup> frequency is greater than Audi. Audible range (i.e 20000 Hz). Noise may damage your hearing and cause other health effects such as it stress it Hyper sensitivity to noise it increase blood pressure it increase Heart rate.

→ Noise produce in mines b from operation of machines, running in vehicle, drilling, blasting, etc. most people are protected from long term damage in a working day (8 hours) by keeping exposure around 85 db but if noise exposure becomes more intentions, damage may occur in a shorter time.

→ Sudden noise level in excess of the Peak exposure standard of 140 db are considered to be hazardous and capable of causing immediately hearing damage.

### Identifying Noise Hazard

→ noise assessment should only be conducted by a competent person and he does the preliminary ~~assess~~ assessment to identify the sources of hazardous noise. Any Assessment should be done in consultation with those who understand the work process such as affected workers their health safety representative

### How to Tackle noise pollution?

→ sources of excess an disruptive noise which make it difficult to hear a normal voice with in 1m of noise sources.

→ workers



How to tackle noise pollution:

→ Keep noise level below the exposure standard of 85db in an 8 hour day for SOP that critical situation can be still be communicate despite noise.

→ substitute noise machinery with ~~that~~ quiet machine.

→ Introduce engineering control to treat noise at its source or in its transmission path such as using sound silencer, noise barrier or partition or isolation

Assessment of noise during blasting:

→ Blasting poses various health problems due to intense noise and vibration created during explosion.

| Blasting limits<br>surface mining    | Sensitive or commercial place<br>criteria  |
|--------------------------------------|--|
| Airblast overpressure                | 115 db peak for 9/10 consecutive blast and not greater than 120 db peak at any time.               |
| Ground vibration & Particle velocity | 5mm/s peak particle velocity for 9/10 consecutive blast and not greater than 10mm/s peak particle. |

### Accumulation Test

→ It is done with normal flame.

→ Accumulation test is done before percentage test to detect percentage of  $\text{CH}_4$  above 4%.

→ In accumulation test the length of flame rises suddenly by the burning of combustible gas mixture burn inside the chamber.

→ The accuracy in accumulation test can't be achieved when the percentage of  $\text{CH}_4$  lies between 1.5% - 4%.

→ when flame sprays or jumps the percentage of gas taken nearly 3% or more.

### Percentage Test

→ but it is done with testing flame.

→ while percentage test done after accumulation test if the percentage of  $\text{CH}_4$  is below 4%.

→ while in percentage test the length of testing flame rises gradually with the increase of percentage of  $\text{CH}_4$ .

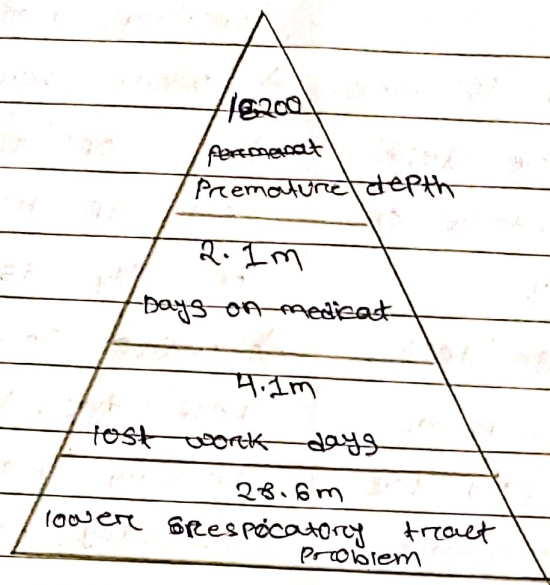
→ The accuracy in percentage test can be achieved when the percentage of  $\text{CH}_4$  lies between 1.5 - 4%.

→ It is unnecessary to conduct the percentage

test when the flame sprays or in a safety lamp as it is declared that the gas percentage is not less than 3%.

## Occupational diseases in mines:- (IMP)

Cases with lower respiratory tract problems:-



## Pneumoconiosis :-

→ The alveoli and pulmonary lymphnode are mostly affected by dust particles.

## Silicosis :-

→ Silicosis is often fatal lung disease caused by breathing dust containing crystalline silica particles. A basic component of sand and Gypsum.

→ There is no cure for silicosis.

## Symptoms :-

→ Shortness of breath

→ Fever

→ Bluish skin at the ear lobe and lips

As the disease progresses it makes fatigue of lungs, extreme shortness, loss of appetite, chest pain and respiratory failure.

### Asbestosis :-

→ Asbestos fibers get into the organ while breathing are  
 ↳ Esophagus ↳ Pleural membrane  
 ↳ Diaphragm ↳ Abdomen ↳ Heart ↳ Larynx

### Black lung Returns :-

→ This is one of the most serious diseases facing by workers in coal industry.

→ The mining boom and new longwall mining technique have exposed more miners to more dust.

### Vibration :-

→ Whole body vibration occurs from operating large mobile equipment  
 drillers, hammer, pile driver, tractor, excavator etc.

## Mine Illumination

### Illumination

→ Remove the darkness is known as illumination

### Severity Index:

→ It is a parameter to check the proness to accident in mines.

$$S.I > 12$$

$$S.I = \frac{(50F + S) \times 100000}{\text{man shift worked}}$$

F = fatal accident

S = Serious to bodily injured

Q) Accident records of a coal mine reveals that 3 fatality and 8 serious injury in 2003. with the total man shift worked during the same period was 390000.

ans :-

$$S.I = \frac{(50F + S) \times 100000}{\text{man shift worked}}$$

$$= \frac{(50 \times 3 + 8) \times 100000}{390000}$$

$$= \frac{158 \times 100000}{390000}$$

$$= \frac{1580}{39}$$

$$= 40.51$$

Q) An underground coal mining employing 1200 person experienced roof falling in which resultant 12 roof fall injury during 2005. The injury per 1000 person employing during 2005.

ans: 1200 person 12 roof falling

$$1 \text{ person roof falling} = 12/1200$$

$$1000 \text{ person roof falling} = \frac{12}{1200} \times 1000$$

$$= 10$$

Mine Rescue:-

Accumulation and accident data analysis:-

Accident:-

Bad event:-

→ frequency rate (employment basis)

$$\frac{\text{no. of person involve in accident} \times 1000}{\text{total employe worked}}$$

→ frequency rate (production rate)

$$\frac{\text{no. of person involve in accident} \times 1000}{\text{Total production}}$$

Q) what is the frequency rate of accident of a mine where employing of 5000 person and there was a total injury, 3 reportable injury and 3 miner injury.

ans:-

$$\frac{\text{no. of accident} \times 1000}{\text{total employe worked}}$$

$$= \frac{3}{5000} \times 1000 = \frac{3}{5} = 1.8$$

Q) 500 coal miners were randomly selected from an underground coal mine. It was found that 50 workers experienced injury in the year 2014. The contribution of a injury based on younger age group (age  $\leq$  40 years) and old age group (age  $>$  40 years)

| Age   | Injured | Non-injured | Total |
|-------|---------|-------------|-------|
| Young | 120 (A) | 130 (B)     | 150   |
| Old   | 80 (C)  | 320 (D)     | 350   |
|       | 50      | 450         | 500   |

The odd injury for younger age group is compared to other age group.

Ans: Odds Ratio =  $\frac{A \times D}{B \times C} = \frac{120 \times 320}{130 \times 80} = \frac{64}{39} = 1.64$

Relative Risk =  $\frac{\frac{A}{A+B}}{\frac{C}{C+D}} = \frac{20}{30} \times \frac{150}{350} = \frac{10}{45}$

21.55

Attributed Risk =  $\frac{A}{A+B} - \frac{C}{C+D} = \frac{20}{150} - \frac{30}{350}$

=  $\frac{2}{15} - \frac{3}{35}$

=  $\frac{14-9}{105}$

=  $\frac{5}{105}$

=  $\frac{1}{21}$

### Illumination

→ Illumination is the providing of light to the darkness area to enable worker to visualize equipments and sites where there worker.

→ It is a illuminance flux ( $\Phi$ ) per unit area.

$$E = \frac{\Phi}{A}$$

Unit of illuminance flux = lumen

Unit of illumination = lux

$$1 \text{ lux} = \frac{1 \text{ lumen}}{1 \text{ m}^2}$$

### Illumination intensity:

$$I = \frac{\Phi}{\omega} \text{ (solid angle)}$$

→ Unit of illumination intensity is candle

$$\text{solid angle } (\omega) = \frac{\text{area}}{(\text{radius})^2}$$

→ Unit of solid angle is steradian

### Mean spherical candle power:

→ It is the average candle power of a lamp in all direction.

$$\text{MSCP} = \frac{\text{flux}}{\text{solid angle}} = \frac{\text{lumen}}{\omega}$$

$$\text{MSCP (I)} = \frac{\text{lumen}}{4\pi} \text{ } (\because \text{ spherical of solid angle } 4\pi)$$

$$\text{I lumen} = \text{MSCP} \times 4\pi$$

→ measurement of solid angle is Goniometer.



Laws of Illumination : (V.V.I)Laws-1

→ According to this law illumination of a 1st law is illumination is inversely proportional to the square of distance between the source and surface.

$$E = \frac{I}{d^2}$$

Laws-2

→ Illumination is directly proportional to the cosine between the normal to the surface and direction of incident light.

$$E = \frac{I}{d^2} \times \cos \theta$$

## Mine Rescue and Recovery

### Proto mark IV apparatus:

→ It consist of

a) A light alloy hollow cylinder of 2 liter capacity containing 300 liter of oxygen compressed to 150 kg/cm<sup>2</sup>. It is fitted with main valve the pressure gauge valve, A bypass valve A reducing valve. The main valve is the cylinder closing and opening valve which is kept open by locking device when the apparatus is in use. The reducing valve the reduces of the pressure oxygen supplied to the wearer and ensure 2 liter oxygen per minute. The Bypass valve is manually operated by the wearer if the reducing valve fail or when the wearer needs more oxygen then the supplied by the reducing valve. The pressure gauge valve admits high pressure oxygen to the pressure gauge.

b) Breathing bag is made of vulcanised rubber and divided into 2 compartment by a partition. The bag contains 2 kg of CO<sub>2</sub> absorbent known as Protosorb. It is the mixture of calcium hydroxide and caustic soda, it keeps the percentage of CO<sub>2</sub> in the breathing circuit below 2%.

d) A cooling chamber of copper containing sodium phosphate which is an crystal form an ordinary temperature.

d) Inhalation and exhalation valve and relief valve. The relief valve allows the escape of any oxygen in excess of wearers requirement.

e) Noseclip, mouthpiece inhalation and exhalation tube.